

SES Microbial Methods

Winogradsky Column

6 Sept 2018

This lab will be used to construct a Winogradsky column from sediments and water collected from Little Sippewisset Marsh, MA. Since it takes several weeks to months for the microbial communities to develop in these columns, the lab and lectures describing the microbial processes at work will occur in Lab/Lecture 5 (week 5/6) to allow time for the community to develop.

A Winogradsky column, named after the Russian microbiologist Sergei Winogradsky, is easily constructed from lake, river, or estuarine water and sediments. As oxygen is depleted in the column by microbial respiration, microenvironments arise which foster the development of a diverse community of microorganisms that are able to catalyze extraordinary biogeochemical reactions. These columns will be used to demonstrate photoautotrophic, chemoheterotrophic, photoheterotrophic, and chemoautotrophic growth. In particular, bacteria capable of sulfate reduction, anaerobic photosynthesis, fermentation, sulfide/sulfur oxidation, iron oxidation, and methanogenesis will develop and flourish within the column. We will also use these columns as a surrogate for field-based sampling during the course.

Materials (Supplied)

- Winogradsky column, ~1400 mL, with 5 sample ports.
- Carbon sources: cotton (cellulose), wool (keratin protein) or chitin (crushed lobster shells)
- Inorganic compounds: CaSO_4 , CaCO_3 , NH_4Cl , PO_4
- Bucket

Field Site: Little Sippewisset marsh, Falmouth.

Procedure:

- Collect enough **sandy** sediments from Little Sippewisset estuary to fill ALL saltwater columns about half way (~700 mL ea) and place your sediments in ONE large container with the other student's and mix. Do not use muddy sediments, as they will plug sample ports.
- Remove ~700 mL of sediments from above and place in separate beaker with carbon source and inorganic compounds (see Treatments for amounts).
- Mix sediments and additives with enough water to produce slurry and pour/scoop into column. Shake or mix sediments to remove large air pockets from column. Sediment-water interface should be just above port #2 from top.
- Collect water sample in 1 l bottle to add to column when back at lab.
- At lab, top off column with collected water and place in light (except for dark treatments).

- Over the next five weeks, observe color development and smell. Are the anaerobic photosynthesizers visible? Can you identify the location of the aerobic-anaerobic interface?

Treatments:

1. Control, no additions
2. 15 g cotton, 5 g CaSO₄, 5 g CaCO₃, 25 mg NH₄Cl, 10 mg K₂HPO₄
3. 15 g cotton, 5 g CaSO₄, 5 g CaCO₃, 25 mg NH₄Cl, 10 mg K₂HPO₄
4. Same as 2., but place in the dark.
5. Same as 2., but place in the dark.
6. 15 g chitin, 5 g CaSO₄, 5 g CaCO₃, 25 mg NH₄Cl, 10 mg K₂HPO₄
7. 15 g chitin, 5 g CaSO₄, 5 g CaCO₃, 25 mg NH₄Cl, 10 mg K₂HPO₄
8. 15 g wool, 5 g CaSO₄, 5 g CaCO₃, 25 mg NH₄Cl, 10 mg K₂HPO₄
9. 15 g wool, 5 g CaSO₄, 5 g CaCO₃, 25 mg NH₄Cl, 10 mg K₂HPO₄
10. *Freshwater*. Control, no additions
11. *Freshwater*. 15 g cotton, 5 g CaCO₃, 25 mg NH₄Cl, 10 mg K₂HPO₄
12. *Freshwater*. 15 g cotton, 5 g CaCO₃, 25 mg NH₄Cl, 10 mg K₂HPO₄

Although we will be covering this experiment in week 6, you might want to look at the following web site that describes Winogradsky columns:

<http://www.biology.ed.ac.uk/research/groups/jdeacon/microbes/winograd.htm>

Some results from a previous year's class can be found at:

<http://ecosystems.mbl.edu/SES/MicrobialMethods/Winogradsky/default.htm>